

15666

Porphyritic Pigeonite Basalt

10.1 grams



Figure 1: Photo of 15666. Cube is 1 cm. S71-49757.

Mineralogical Mode

Olivine	2 %
Pyroxene	40
Plagioclase	
Opaques	5
Silica	
Meostasis	53
Dowty et al. 1973	

Introduction

15666 was collected as part of the large rake sample from station 9a, on the rim of Hadley Rille. It is a pyroxene-phyric basalt with a variolitic groundmass (figure 2). It also includes some olivine.

Petrography

Dowty et al. (1973, 1974) and Nehru et al. (1974) studied the pyroxene phenocrysts (figure 3). They are euhedral, elongate and chemically zoned, with distinct boundaries. Vesicles and metallic iron grains are present. The groundmass is finely crystalline.

15666 was rapidly cooled. Using controlled experiments, Lofgren et al. (1974, 1975) and Grove and Walker (1977) determined the cooling rate and concluded that the rock formed about 15 cm from a "conductive boundary".

Chemistry

Ma et al. (1976) give an analysis.

Processing

There are two thin sections of 15666.

References for 15666

Butler P. (1971) Lunar Sample Catalog, Apollo 15. Curators' Office, MSC 03209

Dowty E., Conrad G.H., Green J.A., Hlava P.F., Keil K., Moore R.B., Nehru C.E. and Prinz M. (1973a) Catalog of Apollo 15 rake samples from stations 2 (St. George), 7 (Spur Crater) and 9a (Hadley Rille). Inst. Meteoritics Spec. Publ. No 11, 51-73. Univ. New Mex. ABQ.

Dowty E., Prinz M. and Keil K. (1973b) Composition, mineralogy, and petrology of 28 mare basalts from Apollo 15 rake samples. *Proc. 4th Lunar Sci. Conf.* 423-444.

Grove T.L. and Walker D. (1977) Cooling histories of Apollo 15 quartz-normative basalts. *Proc. 8th Lunar Sci. Conf.* 1501-1520.

Lofgren G.E., Donaldson C.H. and Usselman T.M. (1975) Geology, petrology and crystallization of Apollo 15 quartz-normative basalts. *Proc. 6th Lunar Sci. Conf.* 79-99.

LSPET (1972a) The Apollo 15 lunar samples: A preliminary description. *Science* **175**, 363-375.

LSPET (1972b) Preliminary examination of lunar samples. Apollo 15 Preliminary Science Report. NASA SP-289, 6-1—6-28.

Ma M.-S., Murali A.V. and Schmitt R.A. (1976) Chemical constraints for mare basalt genesis. *Proc. 7th Lunar Sci. Conf.* 1673-1695.

Nehru C.E., Prinz M., Dowty E. and Keil K. (1974) Spinel-group minerals and ilmenite in Apollo 15 rake samples. *Am. Mineral.* **59**, 1220-1235.

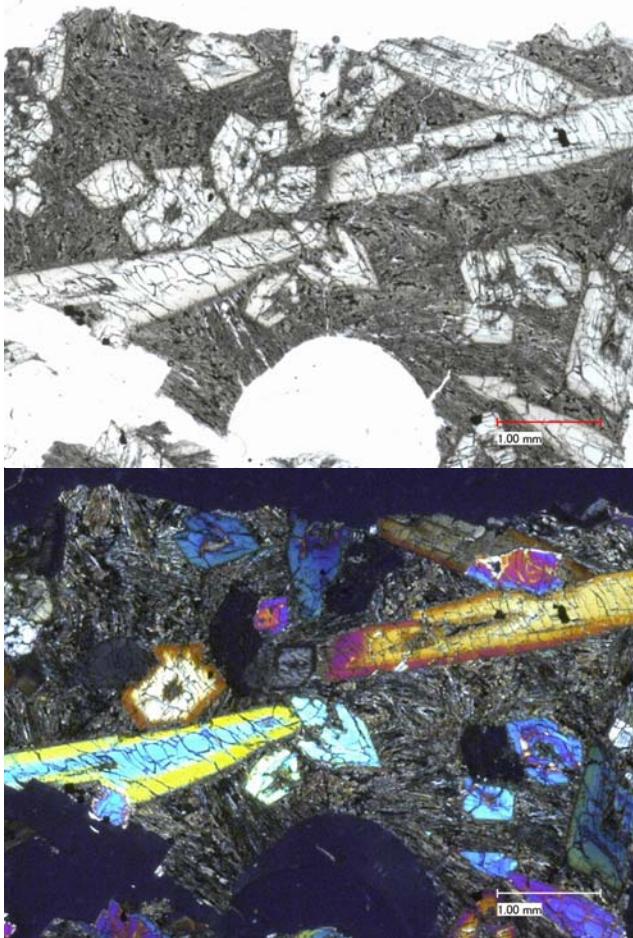


Figure 2: Photomicrographs of thin section 15666,8 by C Meyer @50x.

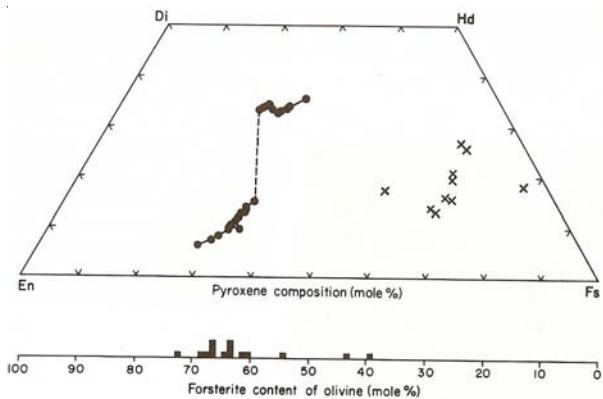


Figure 3 : Pyroxene and olivine composition of 15666 (Dowty et al. 1973)

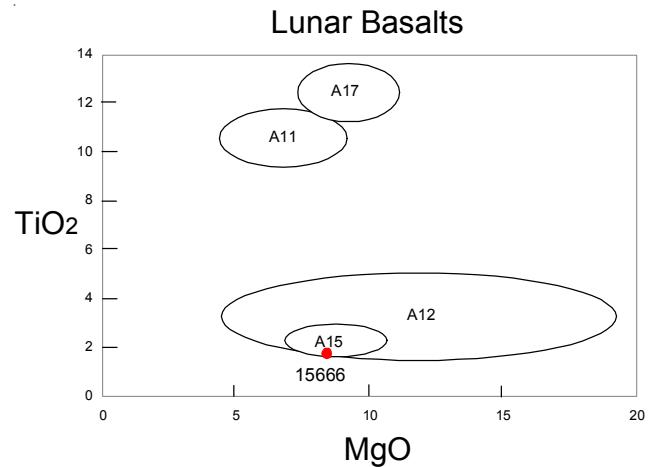


Figure 4: Chemical composition of 15666 compared with other Apollo basalts.

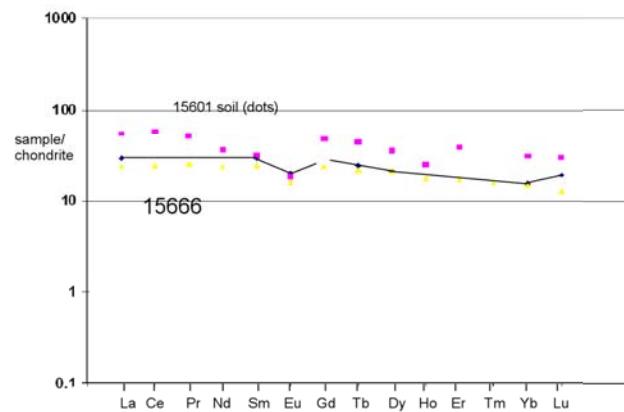


Figure 5: Normalized rare-earth-element diagram for 15666, with 15601 soil for comparison.

Ryder G. (1985) Catalog of Apollo 15 Rocks (three volumes). Curatorial Branch Pub. # 72, JSC#20787

Swann G.A., Hait M.H., Schaber G.C., Freeman V.L., Ulrich G.E., Wolfe E.W., Reed V.S. and Sutton R.L. (1971b) Preliminary description of Apollo 15 sample environments. U.S.G.S. Interagency report: 36. pp219 with maps

Swann G.A., Bailey N.G., Batson R.M., Freeman V.L., Hait M.H., Head J.W., Holt H.E., Howard K.A., Irwin J.B., Larson K.B., Muehlberger W.R., Reed V.S., Rennilson J.J., Schaber G.G., Scott D.R., Silver L.T., Sutton R.L., Ulrich G.E., Wilshire H.G. and Wolfe E.W. (1972) 5. Preliminary Geologic Investigation of the Apollo 15 landing site. In Apollo 15 Preliminary Science Rpt. NASA SP-289. pages 5-1-112.

Table 1. Chemical composition of 15666.

reference	Ma76	Dowty73
weight		
SiO ₂ %	46.9	(b)
TiO ₂	2.3	(a) 1.97 (b)
Al ₂ O ₃	10.3	(a) 9.2 (b)
FeO	21.3	(a) 21.3 (b)
MnO	0.265	(a) (b)
MgO	7.2	(a) 9.5 (b)
CaO	10.2	(a) 9.7 (b)
Na ₂ O	0.372	(a) 0.37 (b)
K ₂ O	0.063	(a) 0.02 (b)
P ₂ O ₅		0.08 (b)
S %		
sum		
Sc ppm	42	(a)
V	176	(a)
Cr	3320	(a) 3015 (b)
Co	37	(a)
Ni	49	(a)
Cu		
Zn		
Ga		
Ge ppb		
As		
Se		
Rb		
Sr		
Y		
Zr		
Nb		
Mo		
Ru		
Rh		
Pd ppb		
Ag ppb		
Cd ppb		
In ppb		
Sn ppb		
Sb ppb		
Te ppb		
Cs ppm		
Ba	40	(a)
La	6.8	(a)
Ce		
Pr		
Nd		
Sm	4.3	(a)
Eu	1.12	(a)
Gd		
Tb	0.88	(a)
Dy	5.1	(a)
Ho		
Er		
Tm		
Yb	2.5	(a)
Lu	0.47	(a)
Hf	3.2	(a)
Ta		
W ppb		
Re ppb		
Os ppb		
Ir ppb		
Pt ppb		
Au ppb		
Th ppm	0.41	(a)
U ppm		
technique	(a) INAA, AA (b) broad-beam e-probe	

